

Research Article**Microwave-assisted one pot synthesis of fused [1,2,3]triazolo-pyrano[3,2-*h*]quinolines and their biological evaluation**Ramesh Babu H.^a, M. Ravinder^b, Sirassu Narsimha^{b*}^aDepartment of Physical Sciences/Chemistry, Kakatiya Institute of Technology and Science, Warangal, T. S-506015, India^bDepartment of Chemistry, Chaitanya Degree College (Autonomous), Warangal, TS-506 001, India

Received: 28 July 2019

Revised: 22 August 2019

Accepted: 27 August 2019

Abstract

Objectives: Microwave-assisted one-pot synthesis of fused [1,2,3]triazoly[4',5':4,5]pyrano[3,2-*h*]quinolines and evaluation of their anticancer activity. **Materials and Methods:** The newly synthesized compounds were characterized by spectroscopic (FTIR, ¹H NMR, ¹³C NMR and Mass) analysis after synthesis. All compounds were screened for their *in vitro* cytotoxic activity against MCF-7, A-549, and HeLa tumor cell lines using MTT analysis. **Results:** The results of cytotoxic activity revealed that compound **3j** has shown potent activity against MCF-7 and HeLa with IC₅₀ values 16.88 ± 0.5 & 11.42 ± 0.6 μM, which are comparable to the standard drug, doxorubicin. **Conclusions:** We have developed a new method for the synthesis of fused 1,2,3-triazole derivatives and the protocol involves a Cu-catalyzed one pot [3+2] cycloaddition followed by C-C bond coupling reaction under microwave condition. Most of the analogues showed strong activity against HeLa and moderate activity against MCF-7 and A-549. The results indicate that these compounds have the potential to develop as leads, and their additional simple structural modifications in the title compounds can produce promising anti-cancer agents for the human cervical cancer HeLa cell line.

Keywords: Microwave synthesis, fused [1,2,3]triazole, anticancer activity

Introduction

1,2,3-Triazoles are an important class of heterocycles that exhibit a wide range of structural and biological activities and are widely used in organic, medicinal, and materials science (Fournier et al., 2007; Hartmuth and Sharpless, 2003; Michael et al., 2008; Alvarez et al., 1994; Tienanet al., 2004; Angell and Burgess, 2007). As compounds containing fused 1,2,3-triazoles become increasingly common in anticancer active substances (Sheng-Jiao et al., 2010; Chung-Yu et al., 2013; Kishna et al., 2015; Narsimha et al., 2016a, 2018b) (**Figure 1**). Based on their structural and biological importance of fused 1,2,3-triazoles, new strategies to synthesize this class of molecules are highly desirable. There are numerous methods available for the synthesis of 1, 4, 5-trisubstituted 1,2,3-triazoles. A classical method involves C-C bond formation of 1, 4-disubstituted 1,2,3-triazole with aryl halide using palladium or copper catalysts in high temperature and long time refluxing condition

(Ackermann et al., 2008, 2009, 2010; Panteleev et al., 2010; Rajkumar et al., 2012).

On the other hand, the quinoline nucleus is an important scaffold found in a variety of biologically active compounds, including natural products and synthetic drugs (Musiol et al., 2007, 2010; Michael, 2007; Kaur et al., 2010). Quinoline and its derivatives have been described as antibacterial (Desai et al., 2012), anticancer (Atwell et al., 1989) and antimalarial (Insuasty et al., 2013). The association of quinoline with the 1,2,3-triazole ring has recently led to the synthesis of new antimicrobial and antitubercular drugs (Sumangala et al., 2010; Thomas et al., 2010, 2011; Harjinder et al., 2014) (**Figure 2**).

Based on the above successful synthesis of fused 1,2,3-triazole derivatives via intramolecular C-H arylation of *in situ* generated 1,4-disubstituted 1,2,3-triazoles and previous therapeutic properties of triazoles and quinolines, as well as continuation of our research on 1,2,3-triazoles (Narsimha et al., 2014, 2016b, 2018a; Ranjith et al., 2017; Swamy et al., 2016, 2017, 2017a; Vasudeva Reedy et al., 2016a, 2016b), we report an efficient method for the synthesis of fused heterocyclic derivatives having triazole and quinoline

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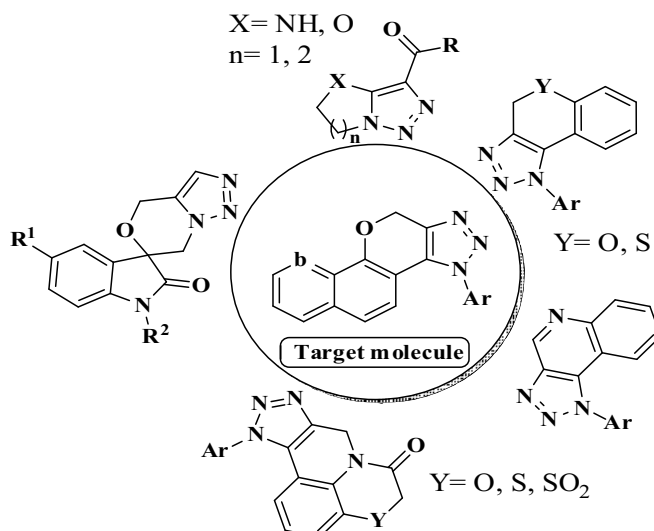


Figure 1. Representative examples of anticancer active fused 1,2,3-triazole moieties

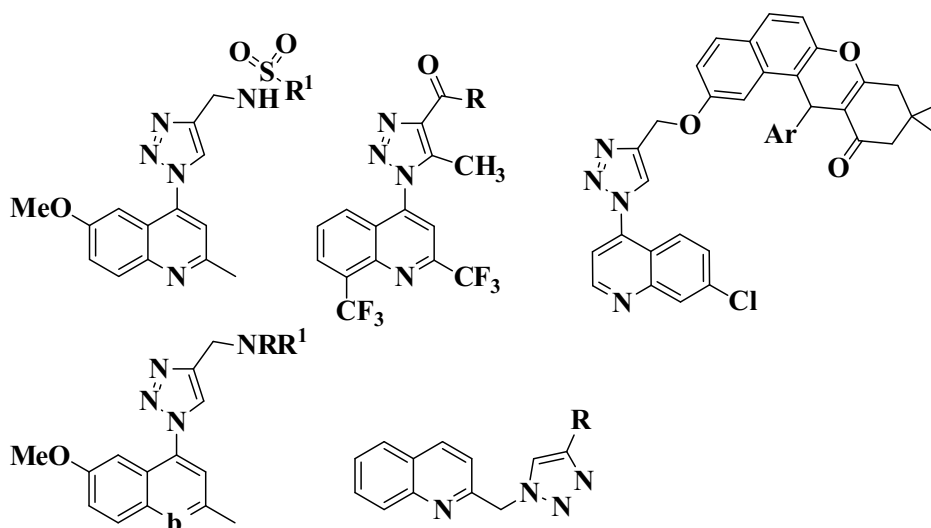
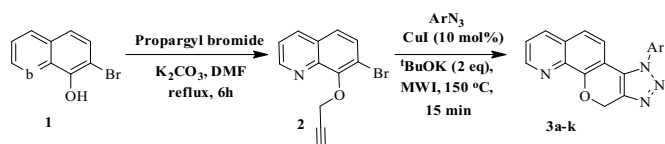


Figure 2. Representative examples of bioactive quinolinecontaining 1,2,3-triazole moieties

moieties in a single scaffold by using copper and palladium catalyzed intramolecular C–H arylation of *in situ* generated 1,4-disubstituted 1,2,3-triazoles from 7-bromo-8-(prop-2-ynoxy) quinoline with different aryl azides in one pot microwave irradiation method and evaluated for their anticancer activity (Scheme 1).



Scheme 1. Synthesis of novel fused [1,2,3]triazolyl[4',5':4,5]pyrano[3,2-h]quinolines

Materials and methods

All the reagents and solvents were purchased from

Aldrich/Merck and used without further purifications. Thin-layer chromatography (TLC) was performed using Merck silica gel 60 F254 precoated plates (0.25 mm) and Silica gel (100-200 mesh) was used for column chromatography. The progress of the reactions as well as purity of the compounds was monitored by thin layer chromatography with using ethylacetate /hexane (7/3) as eluent. Melting points were determined using a Cintex apparatus and are uncorrected. IR spectra (KBr pellet) were recorded on a Perkin-Elmer BX series FT-IR spectrometer. 400 MHz and 100 MHz NMR spectrometers were used to get ¹H-NMR and ¹³C NMR spectra respectively. Coupling constant (*J*) values are presented in Hertz, spin multiples are given as s (singlet), d (doublet), t (triplet), and m (multiplet). Mass spectra were recorded by using ESI-MS method.

Synthesis and Spectral data

Synthesis of 7-bromo-8-(prop-2-ynoxy) quinoline (2)

To a mixture of 7-bromoquinolin-8-ol (0.02 mol) and K_2CO_3 (0.05 mol) in DMF (50 ml) was added propargylbromide (0.025 mol) and stirred at room temperature for 6h. After completion of the reaction by TLC analysis, the reaction mixture was poured carefully into ice-cold water (25 mL) and the product was extracted with ethyl acetate (2 x 50 mL). The combined organic layer was dried over anhydrous Na_2SO_4 . After filtration, the solvent was evaporated under vacuum and the crude product obtained was purified by column chromatography to afford the pure desired product (86%). Yellow solid. 1H NMR ($CDCl_3$, 400 MHz): δ 8.72 (d, $J=8.6$ Hz, 1H), 8.32 (d, $J=8.0$ Hz, 1H), 7.84 (d, $J=8.4$ Hz, 1H), 7.69-7.74 (m, 1H), 7.42 (d, $J=8.4$ Hz, 1H), 4.89 (d, $J=2.2$ Hz, O-CH₂, 2H), 2.77 (t, $J=2.5$ Hz, 1H, CH); ESI-MS: 262 [M+2H].

Synthesis of 3-aryl-3,11-dihydro-[1,2,3]triazolo[4',5':4,5]pyrano [3,2-h]quinoline (3a-k)

7-bromo-8-(prop-2-ynoxy) quinoline (1mmol), aryl azide (1.2mmol) and $tBuOK$ (2mmol) were suspended in DMF (10 mL) in a 25-mL glass vial equipped with a small magnetic stirring bar. To this was added copper(I) iodide (10mol %), and the vial was tightly sealed with an aluminum/Teflon crimp top. The mixture was heated at 150°C for 30 min. Progress of the reaction was monitored by TLC. Then, the vial was cooled to room temperature and the reaction mixture was poured carefully into ice-cold water (10 mL) and the product was extracted with ethyl acetate (2 x 15 mL). The combined organic layer was dried over anhydrous Na_2SO_4 . After filtration, the solvent was evaporated under vacuum and the crude product obtained was purified by column chromatography (hexane/ethyl acetate gradient) to afford the pure desired product.

3-phenyl-3,11-dihydro-[1,2,3]triazolo[4',5':4,5]pyrano[3,2-h]quinoline (3a): pale red solid; IR (KBr, cm^{-1}) ν_{max} 3074 (Ar-H), 1638 (C=N), 1567 (C=C); 1H NMR (400 MHz, $CDCl_3$): δ_H 8.89 (d, $J=8.6$ Hz, 1H), 8.35 (d, $J=8.0$ Hz, 1H), 7.72-7.83 (m, 3H), 7.60-7.67 (m, 2H), 7.47-7.53 (m, 2H), 7.27-7.36 (m, 1H), 5.62 (s, 2H, O-CH₂); ^{13}C -NMR (100 MHz, $CDCl_3$): δ_C 154.3, 153.2, 144.4, 139.3, 133.4, 131.1, 130.3, 128.9, 124.6, 122.5, 121.6, 120.4, 118.7, 61.7; ESI-MS: 301 [M+H]⁺; Anal. Calcd for $C_{18}H_{12}N_4O$: C, 71.99; H, 4.03; N, 18.66. Found: C, 71.86; H, 4.11; N, 18.79.

3-(4-methoxyphenyl)-3,11-dihydro-[1,2,3]triazolo[4',5':4,5]pyrano[3,2-h]quinoline (3b): White solid; IR (KBr, cm^{-1}) ν_{max} 3074 (Ar-H), 1638 (C=N), 1567 (C=C); 1H NMR (400 MHz, $CDCl_3$): δ_H 8.75 (d, $J=8.2$ Hz, 1H), 8.32 (d, $J=8.0$ Hz, 1H), 7.69-7.78 (m, 1H), 7.68 (d, $J=8.6$ Hz, 2H), 7.46-7.53 (m, 1H), 7.27-7.31 (m, 1H), 7.02

(d, $J=8.6$ Hz, 2H), 5.65 (s, 2H, O-CH₂), 3.86 (s, -OCH₃, 3H); ^{13}C -NMR (100 MHz, $CDCl_3$): δ_C 158.1, 153.7, 144.1, 138.9, 133.5, 132.2, 131.0, 127.8, 124.1, 121.9, 121.2, 119.8, 118.2, 61.5, 57.6; ESI-MS: 331 [M+H]⁺; Anal. Calcd for $C_{19}H_{14}N_4O_2$: C, 69.08; H, 4.27; N, 16.96. Found: C, 68.09; H, 4.19; N, 17.04.

3-(3-methoxyphenyl)-3,11-dihydro-[1,2,3]triazolo[4',5':4,5]pyrano[3,2-h]quinoline (3c): White solid; IR (KBr, cm^{-1}) ν_{max} 3074 (Ar-H), 1638 (C=N), 1567 (C=C); 1H NMR (400 MHz, $CDCl_3$): δ_H 8.73 (d, $J=8.2$ Hz, 1H), 8.31 (d, $J=7.8$ Hz, 1H), 7.76-7.84 (m, 2H), 7.56-7.62 (m, 2H), 7.34-7.42 (m, 2H), 7.20 (s, 1H), 5.62 (s, 2H, O-CH₂), 3.88 (s, -OCH₃, 3H); ^{13}C -NMR (100 MHz, $CDCl_3$): δ_C 157.1, 153.8, 144.6, 138.2, 134.1, 133.0, 131.4, 128.1, 125.3, 124.6, 123.4, 122.7, 121.3, 120.1, 119.4, 117.6, 61.3, 57.3; ESI-MS: 331 [M+H]⁺; Anal. Calcd for $C_{19}H_{14}N_4O_2$: C, 69.08; H, 4.27; N, 16.96. Found: C, 69.00; H, 4.21; N, 17.01.

3-(4-chlorophenyl)-3,11-dihydro-[1,2,3]triazolo[4',5':4,5]pyrano[3,2-h]quinoline (3d):

Pale yellow solid; IR (KBr, cm^{-1}) ν_{max} 3069 (Ar-H), 1641 (C=N), 1558 (C=C), 1329 (C-Cl); 1H NMR (400 MHz, $CDCl_3$): δ_H 8.86 (d, $J=8.6$ Hz, 1H), 8.30 (d, $J=8.2$ Hz, 1H), 7.75-7.88 (m, 3H), 7.56-7.64 (m, 2H), 7.33-7.43 (m, 2H), 5.65 (s, 2H, O-CH₂); ^{13}C -NMR (100 MHz, $CDCl_3$): δ_C 153.2, 144.4, 137.8, 136.2, 132.4, 131.6, 127.5, 126.0, 123.3, 122.3, 120.9, 120.1, 119.8, 118.6, 116.4, 62.0; ESI-MS: 335 [M+H]⁺; Anal. Calcd for $C_{18}H_{11}ClN_4O$: C, 64.58; H, 3.31; N, 16.74. Found: C, 64.52; H, 3.25; N, 16.87.

3-(3-chlorophenyl)-3,11-dihydro-[1,2,3]triazolo[4',5':4,5]pyrano[3,2-h]quinoline (3e):

Pale yellow solid; IR (KBr, cm^{-1}) ν_{max} 3084 (Ar-H), 1668 (C=N), 1549 (C=C), 1341 (C-Cl); 1H NMR (400 MHz, $CDCl_3$): δ_H 8.82 (d, $J=8.0$ Hz, 1H), 8.33 (d, $J=7.6$ Hz, 1H), 7.88-7.94 (m, 2H), 7.52-7.63 (m, 2H), 7.30-7.42 (m, 3H), 5.62 (s, 2H, O-CH₂); ^{13}C -NMR (100 MHz, $CDCl_3$): δ_C 153.6, 144.6, 138.1, 137.0, 133.6, 132.8, 127.6, 126.4, 124.8, 124.2, 123.6, 122.9, 121.4, 120.4, 119.8, 117.3, 116.9, 61.8; ESI-MS: 335 [M+H]⁺; Anal. Calcd for $C_{18}H_{11}ClN_4O$: C, 64.58; H, 3.31; N, 16.74. Found: C, 64.49; H, 3.22; N, 16.85.

3-(4-nitrophenyl)-3,11-dihydro-[1,2,3]triazolo[4',5':4,5]pyrano[3,2-h]quinoline (3f):

Yellow solid; IR (KBr, cm^{-1}) ν_{max} 3064 (Ar-H), 1653 (C=N), 1555 (C=C), 1351 (NO₂); 1H NMR (400 MHz, $CDCl_3$): δ_H 8.89 (d, $J=8.0$ Hz, 1H), 8.32 (d, $J=7.2$ Hz, 1H), 8.22 (d, $J=2.4$ Hz, 1H), 8.10 (d, $J=8.2$ Hz, 1H), 7.71-7.83 (m, 2H), 7.54-7.69 (m, 2H), 7.34-7.46 (m, 1H), 5.68 (s,

2H, O-CH₂); ¹³C-NMR(100 MHz, CDCl₃): δ_c 153.7, 148.1, 144.7, 138.9, 134.6, 132.7, 131.8, 129.6, 127.8, 125.4, 124.2, 123.8, 121.3, 120.4, 116.7, 62.2; ESI-MS: 346[M+H]⁺; Anal. Calcd for C₁₈H₁₁N₅O₃: C, 62.61; H, 3.21; N, 20.28. Found: C, 62.54; H, 3.14; N, 20.37.

3-(3-nitrophenyl)-3,11-dihydro-[1,2,3]triazolo[4',5':4,5]pyrano[3,2-h]quinoline(3g)

White solid; IR (KBr, cm⁻¹)_{v_{max}} 3074 (Ar-H), 1639 (C=N), 1565(C=C), 1340 (C-NO₂); ¹H NMR (400 MHz, CDCl₃): δ_H 8.83 (d, *J*=8.6 Hz, 1H), 8.35 (d, *J*=8.0 Hz, 1H), 8.21 (s, 1H), 8.07 (m, 1H), 7.76-7.88 (m, 1H), 7.58-7.60 (m, 2H), 7.29-7.56 (m, 2H), 5.69 (s, 2H, O-CH₂); ¹³C-NMR(100 MHz, CDCl₃): δ_c 153.4, 148.3, 144.7, 138.3, 135.0, 132.4, 131.3, 129.4, 127.6, 125.8, 124.7, 124.0, 123.8, 122.2, 121.6, 120.8, 119.2, 62.5; ESI-MS: 346 [M+H]⁺; Anal. Calcd for C₁₈H₁₁N₅O₃: C, 62.61; H, 3.21; N, 20.28. Found: C, 62.53; H, 3.11; N, 20.35.

3-(p-tolyl)-3,11-dihydro-[1,2,3]triazolo[4',5':4,5]pyrano[3,2-h]quinoline(3h)

Pale yellow solid; IR(KBr, cm⁻¹)_{v_{max}} 3062 (Ar-H), 1641 (C=N), 1557(C=C); ¹H NMR (400 MHz, CDCl₃): δ_H 8.80 (d, *J*=8.0 Hz, 1H), 8.27 (d, *J*=7.2 Hz, 1H), 7.72-7.88 (m, 2H), 7.52-7.62 (m, 2H), 7.20-7.50 (m, 3H), 5.62 (s, 2H, O-CH₂), 2.39 (s, 3H, -CH₃); ¹³C-NMR(100 MHz, CDCl₃): δ_c 153.1, 143.5, 138.4, 133.4, 131.3, 127.8, 124.3, 123.4, 122.7, 122.1, 120.8, 120.1, 119.6, 118.7, 61.5, 21.3; ESI-MS: 315[M+H]⁺; Anal. Calcd for C₁₉H₁₄N₄O: C, 72.60; H, 4.49; N, 17.82. Found: C, 72.66; H, 4.42; N, 17.77.

3-(m-tolyl)-3,11-dihydro-[1,2,3]triazolo[4',5':4,5]pyrano[3,2-h]quinoline(3i)

Pale yellow solid; IR (KBr, cm⁻¹)_{v_{max}} 3080 (Ar-H), 1639 (C=N), 1577(C=C); ¹H NMR (400 MHz, CDCl₃): δ_H 8.82 (d, *J*=8.0 Hz, 1H), 8.26(d, *J*=7.8 Hz, 1H), 7.65-7.76 (m, 2H), 7.54-7.62 (m, 2H), 7.42-7.48 (m, 1H), 7.30-7.37 (m, 2H), 5.65 (s, 2H, O-CH₂), 2.43 (s, 3H, -CH₃); ¹³C-NMR(100 MHz, CDCl₃): δ_c 153.5, 143.7, 138.7, 133.6, 132.1, 128.1, 124.7, 123.3, 122.8, 120.6, 120.0, 119.4, 118.6, 61.7, 19.8; ESI-MS: 315 [M+H]⁺; Anal. Calcd for C₁₉H₁₄N₄O: C, 72.60; H, 4.49; N, 17.82. Found: C, 72.69; H, 4.45; N, 17.73.

3-(3,5-dimethylphenyl)-3,11-dihydro-[1,2,3]triazolo[4',5':4,5]pyrano[3,2-h]quinoline(3j) Pale yellow solid; IR (KBr, cm⁻¹)_{v_{max}} 3066 (Ar-H), 1644 (C=N), 1564(C=C); ¹H NMR (400 MHz, CDCl₃): δ_H 8.83 (d, *J*=8.6 Hz, 1H), 8.26 (d, *J*=8.0 Hz, 1H), 7.78-7.86 (m, 2H), 7.49-7.64(m, 2H), 7.44-7.49(m, 2H), 7.35(s, 2H), 7.14(s, 1H), 5.62 (s, 2H, O-CH₂), 2.41 (s, 6H, -CH₃); ¹³C-NMR(100 MHz, CDCl₃): δ_c 153.1, 143.6, 138.4, 131.4, 127.1, 126.7, 125.3, 123.6, 122.4, 121.3, 120.6, 119.8, 118.2, 61.5, 21.1; ESI-MS: 329[M+H]⁺; Anal. Calcd for C₂₀H₁₆N₄O: C, 73.15; H, 4.91; N,

17.06. Found: C, 73.23; H, 4.97; N, 17.01.

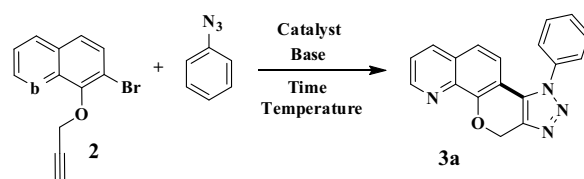
3-(3,5-dichlorophenyl)-3,11-dihydro-[1,2,3]triazolo[4',5':4,5]pyrano[3,2-h]quinoline(3k): Yellow solid; IR (KBr, cm⁻¹)_{v_{max}} 3087 (Ar-H), 1648 (C=N), 1557(C=C), 1327 (C-Cl); ¹H NMR (400 MHz, CDCl₃): δ_H 8.88 (d, *J*=8.0 Hz, 1H), 8.23 (d, *J*=7.2 Hz, 1H), 7.89-8.02 (m, 1H), 7.77(s, 2H), 7.53-7.68(m, 2H), 7.37-7.50 (m, 1H), 5.65 (s, 2H, O-CH₂); ¹³C-NMR(100 MHz, CDCl₃): δ_c 153.8, 144.2 138.9, 134.1, 132.7, 131.2, 126.8, 123.4, 122.3, 121.4, 118.7, 62.2; ESI-MS: 370[M+H]⁺; Anal. Calcd for C₁₈H₁₀Cl₂N₄O: C, 58.56; H, 2.73; N, 15.18. Found: C, 58.48; H, 2.67; N, 15.11.

Cytotoxic activity

All of the synthesized compounds were evaluated for their in vitro cytotoxic activity against three different cancer cell lines such as MCF-7 (breast), A-549 (alveolar) and HeLa (cervical). All the cancer cell lines used in this research work was obtained from National Centre for Cell Sciences (NCCS), Pune, India. Cell viability in the presence of the test samples was measured using the MTT-micro cultured tetrazolium assay (Shekan et al., 1990; Monks et al., 1991). This assay is a quantitative colorimetric method for the determination of cell cytotoxicity. The assessed parameter is the metabolic activity of viable cells. Metabolically active cells reduce pale yellow tetrazolium salt (MTT) to a dark blue water-insoluble formazan, which can be directly quantified after solubilization with DMSO. The absorbance of the formazan directly correlates with the number of viable cells. Human cells were plated into a 96-well plate at a density of 1X10⁴ cells per well. Cells were grown overnight in the full medium and then switched to the low serum media. DMSO was used as a control. After 48 h of treatment with different concentrations of test compounds, the cells were incubated with MTT (2.5 mg mL⁻¹) in the CO₂ chamber for 2 h. The medium was then removed and 100 μL of DMSO was added into each well to dissolve the formazan crystals. After thoroughly mixing, the plates were read at 570 nm for optical density, which is directly correlated with cell quantity. The results were represented as a percentage of cytotoxicity/ viability. All of the experiments were carried out in triplicates. The IC₅₀ values were calculated from the percentage of cytotoxicity and compared with the reference drug.

Results and discussion

The synthetic procedure adopted to obtain the target compounds is shown in **Scheme 1**. The intermediate 7-bromo-8-(prop-2-ynoxy) quinoline (**2**) was prepared with high yields (86%) using 7-bromoquinolin-8-ol (**1**) and propargyl bromide in the presence of anhydrous K₂CO₃ in DMF. With the aim of preparing fused 1, 2, 3-triazoles, our

Table 1. Optimization of CuI-catalyzed intramolecular C-C coupling^a

Entry	Catalyst (10 mol%)	Solvent	Base	Temp (°C)	Time(h)	Yield ^b (%)
1	CuI	THF	<i>t</i> -BuOK	80	20	17
2	CuI	DMSO	<i>t</i> -BuOK	80	20	20
3	CuI	DMF	<i>t</i> -BuOK	80	20	26
4	CuI	THF-H ₂ O	<i>t</i> -BuOK	80	20	16
5	CuI	DMF	<i>t</i> -BuOK	120	24	36
6	CuI	DMF	<i>t</i> -BuOLi	120	24	23
7	CuI	DMF	K ₂ PO ₃	120	24	N.R
8	CuI	DMF	Cs ₂ CO ₃	120	24	N.R
9	CuI	DMF	<i>t</i> -BuOK	140	24	58
10	CuI	DMF	<i>t</i>-BuOK	150W(MW)	15 min	78
11	CuI	DMF	<i>t</i>-BuOK	200W(MW)	20 min	74

^aReactions were performed with **2** (1.0 mmol), azide (1.0 mmol) and *t*-BuOK (2.0 mmol); ^bIsolated yield; N.R. = No Reaction.

initial investigation began with the reaction of **2** and phenyl azide with 10mol % of CuI in THF at 80°C for 20 h. The desired product was formed in 17% yield (**Table 1**, entry 1). Similarly, the reaction was carried out using different solvents DMSO, DMF, and H₂O (**Table 1**, entries 2–4), the product was formed with yields not exceeding 27%, with DMF being the most favorable solvent (26% yield). Then, the reaction was carried out using 10 mol % of CuI at 120°C for 24 h, the desired product was obtained with 36% yield (**Table 1**, entry 5). When *t*-BuOK was replaced with *t*-BuOLi, K₂PO₃ and Cs₂CO₃, we could not get the desired products in good yield compared to *t*-BuOK (**Table 1**, entries 6–8). When the reaction was carried out at higher temperatures (above 140°C) using *t*-BuOK and DMF the product was obtained in 58% yield (**Table 1**, entry 9). Encouraged by these results the reaction was carried out under microwave irradiation, which resulted in the generation of the desired product in excellent yields (**Table 1**, entry 10, 11). After the complete optimization studies, it was clear that the use of catalyst load of CuI (10 mol %) with 2 equivalents of *t*-BuOK in DMF under the microwave irradiation (150 W) was the optimal reaction conditions to obtain the desired products in good yields. Several substituted phenyl azides containing electron-donating groups (methyl, and methoxy groups) and electron-withdrawing groups (chloro and nitro groups), were all compatible with this transformation under the optimal reaction conditions (**Figure 3**).

The structures of the newly synthesized compounds **3a–3k** were

confirmed by spectral data (¹H NMR, ¹³C NMR, IR, and ESI-MS) and elemental (CHN) analysis. All the spectral and analytical data of the synthesized compounds were in full agreement with the proposed structures and also discussed for a representative compound **3b**. From the IR spectrum, the appearance of a broad absorption band at sharp bands at 3074, 1638 and 1567 cm⁻¹ are ascribed to Ar-H, C=N and -C=C stretching frequencies, respectively. From the ¹H NMR spectrum, the signals appearance at 3.86 (s, 3H, O-CH₃), 5.65 (s, 2H, O-CH₂), 7.02 to 8.75 (m, 9H, Ar-H), and from the ¹³C NMR, the presence of signals at 158.1 ppm (C-OCH₃), 153.7 ppm (C-O-CH₂), 61.5 ppm (-OCH₂), 57.6 ppm (-OCH₃) and the ESI- mass spectra of **3b** showed [M+H] ion peak at *m/z* 331, which confirmed the structure of compound **3b**. The elemental analyses (CHN) data (C, 68.09; H, 4.19; N, 17.04) confirmed the purity of compound **3b**.

Anticancer activity

In vitro anticancer activity results revealed that all the compounds were active against all the three tested cell lines such as breast carcinoma (MCF-7), alveolar carcinoma (A-549) and cervical carcinoma (HeLa). Cell viability in the presence of the test samples was measured using the MTT-microcultured tetrazolium assay. Doxorubicin was used as a positive control and the results are summarized in **table 2**.

Among them, the compound derived from the 3,5-dimethylphenyl group on fused triazole ring i.e. **3j** has shown

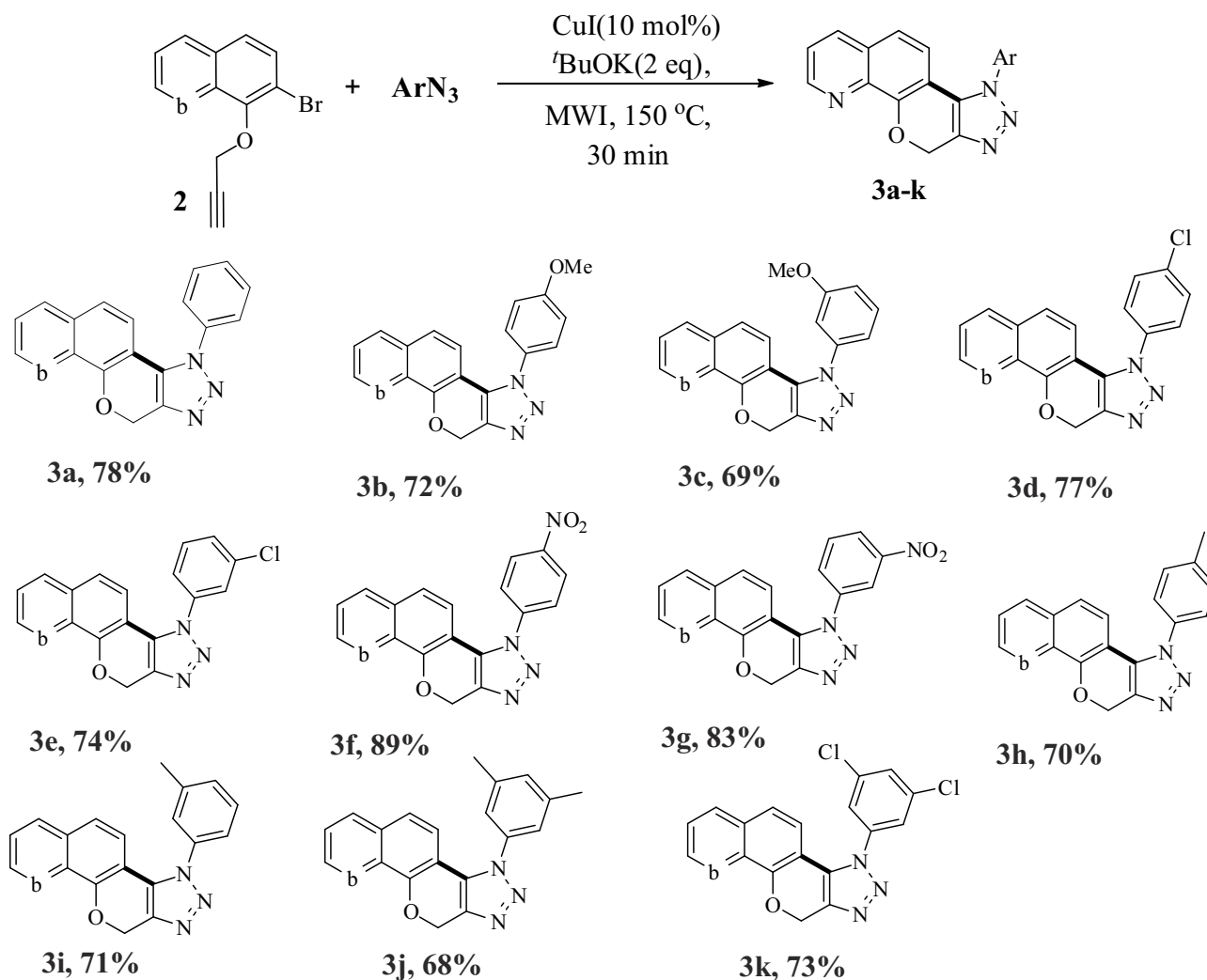


Figure 3. Cu/Pd catalyzed one pot microwave synthesis of fused [1,2,3]triazoly[4',5':4,5] pyrano[3,2-h]quinolines. The isolated yields are given as percentages

Table 2. Cytotoxic activity of fused 1,2,3-triazoles (**3a-3k**) on MCF-7, A-549 and HeLa [*in vitro*^a (IC₅₀μM)]^b

Product	MCF-7	A-549	HeLa
3a	41.86 ± 0.7	62.88 ± 1.2	46.39 ± 1.0
3b	89.21 ± 0.9	77.52 ± 1.3	24.92 ± 0.8
3c	31.44 ± 0.5	44.52 ± 1.1	15.67 ± 0.6
3d	47.67 ± 0.8	59.62 ± 1.1	49.14 ± 1.0
3e	55.27 ± 1.4	75.89 ± 1.6	54.57 ± 0.9
3f	79.88 ± 1.3	96.87 ± 1.5	68.21 ± 1.1
3g	81.37 ± 1.6	69.54 ± 1.0	51.79 ± 0.8
3h	39.18 ± 1.0	42.97 ± 1.3	21.27 ± 0.7
3i	40.66 ± 0.8	63.47 ± 1.1	31.66 ± 0.9
3j	16.88 ± 0.5	32.47 ± 1.0	11.42 ± 0.6
3k	49.55 ± 1.0	67.88 ± 1.4	43.68 ± 1.1
Doxorubicin	3.01 ± 0.2	1.08 ± 0.08	1.35 ± 0.04

^aValues are expressed as mean ± SEM. ^bCytotoxicity,

potent activity against MCF-7 & HeLa with IC_{50} values 16.88 ± 0.5 & $11.42 \pm 0.6 \mu\text{M}$ and moderate activity against A-549 with IC_{50} value $32.47 \pm 1.0 \mu\text{M}$ respectively. Compound **3c**, having a 3-methoxyphenyl group on triazole ring exhibited potent activity against HeLa with IC_{50} value $15.67 \pm 0.6 \mu\text{M}$ and moderate activity against the MCF-7 cell line with an IC_{50} value of $31.44 \pm 0.5 \mu\text{M}$. These results are comparable to those of the standard drug doxorubicin. Similarly, the introduction of 4-methoxyphenyl group and 4-methylphenyl group on fused triazole ring i.e. **3b** and **3h** have shown good activity against HeLa cell line with IC_{50} values 24.92 ± 0.8 & $21.27 \pm 0.7 \mu\text{M}$ respectively. Remaining compounds have shown moderate activity with IC_{50} values ranging from 39.18 ± 1.0 - $96.87 \pm 1.5 \mu\text{M}$ against the three cell lines. It is necessary to point out that all of the potent analogues contain methyl, dimethyl and methoxy substituents on the triazole ring (**3b**, **3c**, **3h**, **3i** & **3j**) and this preference was uniform and irrespective to the substitution pattern on the aromatic ring. The electron withdrawing substituents such as chloro and nitro groups on the triazole ring exhibited moderate to poor activity against all three tested cell lines.

Conclusion

In summary, we have developed a new method for the synthesis of [1,2,3] triazolyl [4', 5': 4,5] pyrano [3,2-h] quinoline derivatives. The protocol involves a Cu-catalyzed one pot [3+2] cyclo addition followed by C-C bond coupling reaction under microwave condition without isolating the intermediate 1,4-disubstituted 1,2,3-triazole. These new compounds were tested for their in vitro cytotoxic activity against the cancer cell lines MCF-7, A-549 and HeLa. Among them, compounds **3c** and **3j** showed strong activity against HeLa. Interestingly, most of the analogues showed strong activity against HeLa and moderate activity against MCF-7 and A-549. In general, the results indicate that these compounds have the potential to develop as leads, and their additional simple structural modifications in the title compounds can produce promising anti-cancer agents for the human cervical cancer HeLa cell line.

Acknowledgments

The authors are thankful to the Director of Indian Institute of Chemical Technology in Hyderabad for recording ^1H , ^{13}C NMR and mass spectra. The authors are thankful to the Head, Department of Bio-Technology, Kakatiya University, and Warangal for providing data of biological activity.

Conflicts of interest

The authors declare no conflict of interest.

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